## **Advanced BODMAS**

# *Rearranging equations with more complicated functions*

### Answers to additional problems

**12.1** The *I* term appears within the denominator of the fraction inside the bracket, which is inconvenient. Using the laws of logarithms, we can rewrite the equation slightly.

$$-\ln\left(\frac{I_0}{I}\right) = -c \,\boldsymbol{\ell} \,\varepsilon \qquad \text{we multiply each side by} -1$$
$$\ln\left(\frac{I}{I_0}\right) = -c \,\boldsymbol{\ell} \,\varepsilon$$

We now take the inverse function of the logarithm,

$$\left(\frac{I}{I_0}\right) = \exp(-c\,\boldsymbol{\ell}\,\varepsilon)$$

Finally, we multiply both sides by  $I_0$ 

$$I = I_0 \exp(-c \ell \varepsilon)$$

**12.2** We first divide both sides by *A* in order to get the exponential on its own

$$\frac{k}{A} = \exp\!\left(-\frac{E_a}{RT}\right)$$

Then take the inverse function of the exponential, which is a natural logarithm ln

$$\ln\left(\frac{k}{A}\right) = -\frac{E_a}{RT}$$

We multiply both sides by -RT

$$-RT\ln\left(\frac{k}{A}\right) = E_a$$

Finally, we can remove the minus sign before the logarithm by inverting the logarithm  $(\ln (a/b) = \ln (b/a)^{-1} = -\ln (b/a))$ 

 $3V = 4\pi r^3$  $\frac{3V}{4\pi} = r^3$ 

۲

$$E_a = RT \ln\left(\frac{A}{k}\right)$$

**12.3 1.** We **MULTIPLY** both sides by 3,

- **2.** We **DIVIDE** both sides by  $4\pi$ ,
- **3.** We take the cube **ROOT** of both sides,
- 4. Only now do we insert terms

$$r = \sqrt[3]{\frac{3V}{4\pi}}$$
$$r = \sqrt[3]{\frac{3 \times 5.56 \times 10^{-31} \text{ m}^3}{4\pi}} \text{ so } r = 5.1 \times 10^{-11} \text{ m}$$

Don't forget to use brackets when dividing by  $4\pi$ .

This follows because  $\left(\frac{I_0}{I}\right) = \left(\frac{I}{I_o}\right)^{-1}$  and  $\log a^n = n \log a$ .

( )

۲





#### 2 12: Advanced BODMAS

**12.4** We first simplify the left-hand side so it comprises only one term, which includes  $\Delta G_{T_2}^{\Phi}$ 

۲

$$\frac{\Delta G_{T_2}^{\oplus}}{T_2} = \frac{\Delta G_{T_1}^{\oplus}}{T_1} + \Delta H^{\oplus} \left(\frac{1}{T_2} - \frac{1}{T_1}\right)$$

We then multiply both sides by  $T_2$ 

$$\Delta G_{T_2}^{\textcircled{red}} = \frac{T_2 \Delta G_{T_1}^{\textcircled{red}}}{T_1} + T_2 \Delta H^{\textcircled{red}} \left(\frac{1}{T_2} - \frac{1}{T_1}\right)$$

The second term on the right-hand side could also be rewritten, so the equation becomes

$$\Delta G_{T_2}^{\oplus} = T_2 \left( \frac{\Delta G_{T_1}^{\oplus}}{T_1} + \Delta H^{\oplus} \left( \frac{1}{T_2} - \frac{1}{T_1} \right) \right) \quad \text{or} \quad \Delta G_{T_2}^{\oplus} = \frac{T_2 \Delta G_{T_1}^{\oplus}}{T_1} + \Delta H^{\oplus} \left( 1 - \frac{T_2}{T_1} \right)$$

**12.5** Looking at the right-hand side of the equation shows how the concentration [I] was first square rooted to form  $\sqrt{[I]}$  which we then MULTIPLIED by *k* [aromatic].

The first step when obtaining [(I)] is therefore to **DIVIDE** both sides of the equation by k [aromatic]. We will treat k [aromatic] as a compound variable. We therefore perform the inverse operation, and **DIVIDE** both sides in a single step,

$$\frac{\text{rate}}{k[\text{aromatic}]} = \sqrt{[\mathbf{I}]}$$

Since the right-hand side is a SQUARE ROOT to obtain [I] from  $\sqrt{[I]}$  we perform the inverse operation and SQUARE both sides,

$$\left(\frac{\text{rate}}{k \text{[aromatic]}}\right)^2 = \left(\sqrt{[\mathbf{I}]}\right)^2$$

The SQUARE of a SQUARE ROOT produces the thing itself. In this case, it produces [I],

$$[\mathbf{I}] = \left(\frac{\text{rate}}{k \,[\text{aromatic}]}\right)^2$$

**12.6** We first make the logarithm the subject, so we subtract k and divide by -a,

$$\log_{10}[\mathbf{F}^-] = \frac{emf - k}{-a}$$

We multiply both top and bottom of the right-hand side by -1 and then take the inverse function of  $\log_{10}$ , which is  $10^{\rm x}$ 

$$[F^-] = 10^{\left(\frac{k-emf}{a}\right)}$$

**12.7** To make *c* the subject of the equation, we first **SUBTRACT**  $\Lambda^{\circ}$  from both sides,

$$\Lambda - \Lambda^{\circ} = -b\sqrt{c}$$

Note how the minus sign persists on the right-hand side. Next, we note how  $\sqrt{c}$  has been MULTIPLIED by a factor of -b, so we DIVIDE both sides by -b,

$$\frac{\Lambda - \Lambda^{\circ}}{-b} = \sqrt{c}$$

Finally we perform the inverse operation to SQUARE ROOT, and square both sides,

$$c = \left(\frac{\Lambda - \Lambda^{\circ}}{-b}\right)^2$$

We could rewrite this result, multiplying top and bottom within the bracket by '-1',

( )

 $(\mathbf{r})$ 

۲

3

 $(\mathbf{r})$ 

$$c = \left(\frac{\Lambda^{\circ} - \Lambda}{b}\right)^2$$

• It might have been easier to rewrite the source equation as  $\Lambda = \Lambda^{\circ} + (-b\sqrt{c})$  to make it clear which term we should isolate first.

**12.8** We first need to simplify the right-hand side of the equation by separating the bracket from the remainder of the equation, getting it on its own. So we cross-multiply

$$-\frac{R}{\Delta H^{\textcircled{o}}}\ln\left(\frac{K_2}{K_1}\right) = \left(\frac{1}{T_2} - \frac{1}{T_1}\right)$$

The bracket on the right-hand side is now redundant

$$-\frac{R}{\Delta H^{\oplus}}\ln\left(\frac{K_2}{K_1}\right) = \frac{1}{T_2} - \frac{1}{T_1}$$

We then move the  $1/T_1$  term

$$\frac{1}{T_2} = \frac{1}{T_1} - \frac{R}{\Delta H^{\odot}} \ln\left(\frac{K_2}{K_1}\right)$$

Finally, we take the reciprocal of both sides,

$$T_2 = \frac{1}{\left(\frac{1}{T_1} - \frac{R}{\Delta H^{\oplus}} \ln\left(\frac{K_2}{K_1}\right)\right)} \text{ so } T_2 = \left(\frac{1}{T_1} - \frac{R}{\Delta H^{\oplus}} \ln\left(\frac{K_2}{K_1}\right)\right)^{-1}$$

**12.9** 1. The sixth **ROOT** has been taken, to form  $\sqrt[6]{\upsilon}$ .

**2.**  $\sqrt[6]{\upsilon}$  has been **DIVIDED** into a large collection of constants  $0.62nFAc D^{\frac{1}{2}}\omega^{\frac{1}{2}}$ . This collection certainly looks formidable, but we do not really need to think about them. If it makes the equation look less scary, rewrite it as  $k \div \sqrt[6]{\upsilon}$ , where *k* represents all these constants bundled together.

To rearrange the equation, we reverse these two operations,

**1.** To obtain  $\sqrt[6]{v}$  on its own, we **MULTIPLY** both sides of the equation by  $\sqrt[6]{v}$  and **DIVIDE** both sides by *I*. We obtain,

$$\sqrt[6]{\upsilon} = \frac{0.62 \, nFA \, c \, D^{\frac{2}{3}} \omega^{\frac{1}{2}}}{I}$$

2. To reverse the sixth **ROOT**, we must take the sixth **POWER**. We obtain,

$$\left(\sqrt[6]{\upsilon}\right)^6 = \left(\frac{0.62 \ nFA \ c \ D^{2/3} \omega^{1/2}}{I}\right)^6$$

The left-hand side then collapses to form

$$v = \left(\frac{0.62 \, nFA \, c \, D^{\frac{2}{3}} \omega^{\frac{1}{2}}}{I}\right)^6$$

**12.10** We first remove the pre-exponential factor by cross-multiplying, and combining the exponentials arguments using the first power law, eqn. (9.4)

$$\frac{kh}{Tk_B} = \exp\left(\frac{\Delta S^{\ddagger}}{R} + \left(-\frac{\Delta H^{\ddagger}}{RT}\right)\right)$$

We then take the inverse function of the exponential which is ln. Doing so deals with both terms on the right-hand side

۲

( )

08-07-2021 13:13:14

#### 12: Advanced BODMAS

$$\ln\!\left(\frac{kh}{Tk_B}\right) = \frac{\Delta S^{\ddagger}}{R} - \frac{\Delta H^{\ddagger}}{RT}$$

Before we go further, for historical reasons, it's usual to split the 'ln' term in a particular way, which we will show in 2 steps

۲

$$\ln\left[\left(\frac{k}{T}\right) \times \left(\frac{k_{B}}{h}\right)^{-1}\right] = \frac{\Delta S^{*}}{R} - \frac{\Delta H^{*}}{RT}$$
$$\ln\left(\frac{k}{T}\right) - \ln\left(\frac{k_{B}}{h}\right) = \frac{\Delta S^{*}}{R} - \frac{\Delta H^{*}}{RT}$$

We next add the enthalpy term to both sides

$$\frac{\Delta S^{\ddagger}}{R} = \ln \frac{k}{T} - \ln \frac{k_B}{h} + \frac{\Delta H^{\ddagger}}{RT}$$

Finally, we multiply both sides by the gas constant

$$\Delta S^{\ddagger} = R \ln \frac{k}{T} - R \ln \frac{k_B}{h} + \frac{\Delta H^{\ddagger}}{T}$$

۲

4

۲